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## The *Pbca* Polymorph of Dichloro( $\eta^4$ -1,5-cyclooctadiene)palladium(II)

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### Abstract

The Pd–C bond length range is 2.189 (2)–2.221 (2) Å, the Pd–Cl lengths are 2.3065 (6) and 2.3072 (6) Å, and the Cl–Pd–Cl angle is 91.84 (2)°. The cyclooctadiene ring is in the twist-boat conformation, with C=C bond lengths of 1.382 (3) and 1.385 (3) Å.

### Comment

The published structure of (1,5-cod)PdCl<sub>2</sub> (cod = cyclooctadiene) has space group *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>, and the crystals obtained from CH<sub>2</sub>Cl<sub>2</sub> are described as both needles and octahedra (Rettig, Wing & Wiger, 1981) or only as ‘un cristal allongé suivant l’axe *c*...’ (the 6.876 Å axis) (Benchekroun, Herpin, Julia & Saussine, 1977). Our crystallizations from CH<sub>2</sub>Cl<sub>2</sub> yielded needles, octahedra and prisms with rhombic cross sections. In order to clarify the uncertainty, we measured the unit-cell dimensions of all three morphological types. We found the needles to have space group *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>, with dimensions *a* = 6.874 (2), *b* = 12.293 (1), *c* = 10.972 (2) Å at 296 K, in good agreement with the published structure. We found

that the rhombic prism and the octahedron are both a second polymorph, of space group *Pbca*, identical to that described by Howells (1973). A rhombic prism yielded unit-cell dimensions *a* = 11.8266 (9), *b* = 11.9875 (12), *c* = 13.0812 (9) Å at 295 K. We have carried out a full structure determination using a flattened octahedron.

The determination reported here represents an increase in precision over the *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub> form by a factor of 2–3, and the agreement between the two molecular structures is good, including the conformation of the cod ring. The root-mean-square deviation between the two sets of eight endocyclic torsion angles is 6.8°, and the largest individual deviation is only 10°, for C2–C3–C4–C5. All four Pd–C bond distances of the *Pbca* form agree with those of the *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub> form within experimental error. While the *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub> form exhibited a difference in Pd–Cl distances of marginal significance, our determination has equal Pd–Cl distances.

The Cl–Pd–Cl angle is slightly larger in the *Pbca* form; 91.84 (2) versus 90.31 (5)°. This angle has a value of 94.02 (5)° in dichloro(norbornadiene)palladium(II) (Baenziger, Richards & Doyle, 1965), 91.11 (3)° in dichloro[(1,2,5,6- $\eta$ )-cyclooctatetraene]palladium(II) (Baenziger, Goebel, Berg & Doyle, 1978), 91.9° in dichloro(1,4-cyclooctadiene)palladium(II) (Rettig *et al.*, 1981) and 90.1° in dichloro(1,5-cyclononadiene)palladium(II) (Rettig *et al.*, 1981).

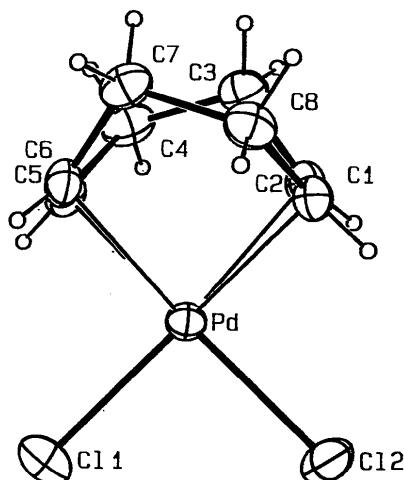


Fig. 1. ORTEP (Johnson, 1976) drawing of the title compound, with thermal ellipsoids drawn at the 40% probability level.

### Experimental

#### Crystal data

[PdCl<sub>2</sub>(C<sub>8</sub>H<sub>12</sub>)]  
*M*<sub>r</sub> = 285.5

Mo K $\alpha$  radiation  
 $\lambda$  = 0.71073 Å

Orthorhombic  
*Pbca*  
 $a = 11.8271$  (4) Å  
 $b = 11.9869$  (4) Å  
 $c = 13.0821$  (5) Å  
 $V = 1854.7$  (2) Å<sup>3</sup>  
 $Z = 8$   
 $D_x = 2.045$  Mg m<sup>-3</sup>

Cell parameters from 25 reflections  
 $\theta = 11\text{--}14^\circ$   
 $\mu = 2.49$  mm<sup>-1</sup>  
 $T = 295$  K  
 Flattened octahedron  
 $0.33 \times 0.32 \times 0.18$  mm  
 Amber

C8—C1—C2—C3      4.8 (3)      C3—C4—C5—C6      37.9 (3)  
 C2—C1—C8—C7      35.3 (3)      C4—C5—C6—C7      4.0 (3)  
 C1—C2—C3—C4      -93.3 (3)      C5—C6—C7—C8      -95.4 (3)  
 C2—C3—C4—C5      38.0 (3)      C6—C7—C8—C1      41.1 (3)

Programs used include *MolEN* (Fair, 1990) and *ORTEPII* (Johnson, 1976).

#### Data collection

Enraf-Nonius CAD-4 diffractometer  
 $\omega\text{-}2\theta$  scans  
 Absorption correction:  
 empirical  
 $T_{\min} = 0.933$ ,  $T_{\max} = 0.998$   
 4544 measured reflections  
 4065 independent reflections

3105 observed reflections  
 $[I > 3\sigma(I)]$   
 $\theta_{\max} = 35^\circ$   
 $h = 0 \rightarrow 19$   
 $k = 0 \rightarrow 19$   
 $l = 0 \rightarrow 21$   
 3 standard reflections  
 frequency: 166.7 min  
 intensity variation: <1%

Lists of structure factors, anisotropic thermal parameters, H-atom coordinates and bond distances and angles involving H atoms have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71258 (30 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: HH1058]

#### Refinement

Refinement on  $F$   
 Final  $R = 0.022$   
 $wR = 0.029$   
 $S = 1.455$   
 3105 reflections  
 149 parameters  
 All H-atom parameters refined  
 $w = 4F_o^2[\sigma^2(I) + (0.02F_o^2)^2]^{-1}$   
 $(\Delta/\sigma)_{\max} = 0.04$

$\Delta\rho_{\max} = 0.44$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.17$  e Å<sup>-3</sup>  
 Extinction correction:  
 $(I + gl_c)^{-1}$  applied to  $F_c$   
 Extinction coefficient:  
 $g = 2.86$  (8)  $\times 10^{-7}$   
 Atomic scattering factors  
 from *International Tables for X-ray Crystallography* (1974, Vol. IV)

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Table 1. Fractional atomic coordinates and equivalent isotropic thermal parameters (Å<sup>2</sup>)

	$x$	$y$	$z$	$B_{\text{eq}}$
Pd	0.35479 (1)	0.60963 (1)	0.69883 (1)	2.029 (2)
C11	0.34127 (5)	0.54235 (5)	0.86362 (4)	3.76 (1)
C12	0.48296 (4)	0.47290 (4)	0.65451 (5)	3.58 (1)
C1	0.4148 (2)	0.7088 (2)	0.5692 (1)	2.54 (3)
C2	0.3195 (2)	0.6521 (2)	0.5371 (1)	2.58 (3)
C3	0.2024 (2)	0.7012 (2)	0.5266 (2)	3.06 (4)
C4	0.1298 (2)	0.6852 (2)	0.6222 (2)	3.13 (4)
C5	0.1932 (2)	0.6972 (2)	0.7201 (2)	2.80 (3)
C6	0.2789 (2)	0.7728 (2)	0.7406 (1)	2.71 (3)
C7	0.3192 (2)	0.8626 (2)	0.6690 (2)	2.85 (3)
C8	0.4216 (2)	0.8262 (2)	0.6071 (2)	2.84 (3)

Table 2. Geometric parameters (Å, °)

Pd—Cl1	2.3072 (6)	C1—C8	1.494 (3)
Pd—Cl2	2.3065 (6)	C2—C3	1.511 (3)
Pd—C1	2.189 (2)	C3—C4	1.528 (3)
Pd—C2	2.216 (2)	C4—C5	1.492 (3)
Pd—C5	2.198 (2)	C5—C6	1.385 (3)
Pd—C6	2.221 (2)	C6—C7	1.504 (3)
C1—C2	1.382 (3)	C7—C8	1.520 (3)
C11—Pd—Cl2	91.84 (2)	C4—C5—C6	126.6 (2)
C2—C1—C8	127.4 (2)	C5—C6—C7	125.5 (2)
C1—C2—C3	125.7 (2)	C6—C7—C8	112.3 (2)
C2—C3—C4	113.0 (2)	C1—C8—C7	113.8 (2)
C3—C4—C5	114.1 (2)		

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#### Bis( $\mu$ -benzoato- $\kappa O:\kappa O'$ )-bis(benzoato- $\kappa O,O'$ )(*N,N*-dimethylformamide- $\kappa O$ )-dioxouranium(VI)]

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#### Abstract

The action of dimethylformamide on diaqua(benzoato)chlorodioxouranium(VI), in the presence of  $\beta$ -cyclodextrin, yields the title compound. The crystal structure of this new complex reveals that the linear uranyl is equatorially surrounded by five O

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